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### POLARIZED SPECTRA OF THE TRIGONAL BIPYRAMIDAL $\text{CoNP}_3\text{Br}$ CHROMOPHORE

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## SHORT COMMUNICATIONS

Polarized Spectra of the Trigonal Bipyramidal  $\text{CoNP}_3\text{Br}$  Chromophore

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The electronic spectra of high-spin pentacoordinate cobalt(II) complexes have been tentatively assigned on the basis of a ligand field approach.<sup>1-3</sup> A controversy has arisen about the assignment of the spectrum of a typical chromophore,  $\text{CoN}_4\text{Br}$ , having strictly  $C_{3v}$  symmetry in the compound  $[\text{CoBr}(\text{Me}_6\text{tren})]\text{Br}$ , where  $\text{Me}_6\text{tren}$  is  $\text{N}[\text{CH}_2\text{CH}_2\text{N}(\text{CH}_3)_2]_3$ . This spectrum shows four distinct bands well separated from charge transfer bands at 5.7, 12.3, 16.1 and 19.2-19.8 kK.<sup>4</sup> Ciampolini and Bertini<sup>2</sup> assigned these bands to transitions from the  $^4A_2(\text{F})$  ground state to the  $^4E(\text{F})$ ,  $^4E(\text{F})$ ,  $^4A_2(\text{P})$ , and  $^4E(\text{P})$  excited states whereas Wood<sup>3</sup> assigned the second band to a spin forbidden transition, the third band to a transition to the  $^4E(\text{F})$  level and the last band to the two transitions to the highest two levels. Because of the cubic symmetry of the crystal lattice of this compound,<sup>5</sup> polarized spectra were of no aid in deciding the correct assignment.

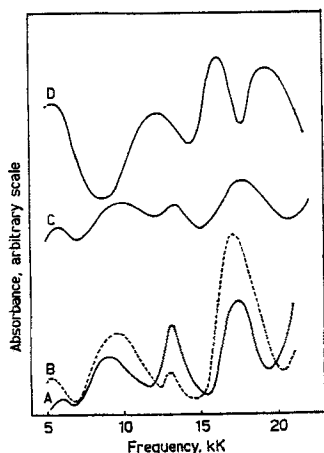


FIGURE 1 Spectra of  $[\text{CoBr}(\text{NP}_3)]\text{PF}_6$ : A, polarized along "a" direction; B, polarized along "b" direction; C, unpolarized reflectance spectrum. The D curve is the reflectance spectrum of  $[\text{CoBr}(\text{Me}_6\text{tren})]\text{Br}$ .

In this communication additional clarifying data are reported concerning the polarized spectra of the chromophore  $\text{CoNP}_3\text{Br}$  in the compound  $[\text{CoBr}(\text{NP}_3)]\text{PF}_6$ , where  $\text{NP}_3$  is  $\text{N}[\text{CH}_2\text{CH}_2\text{P}(\text{C}_6\text{H}_5)_2]_3$ , which has essentially  $C_{3v}$  symmetry.<sup>7,8</sup> The trigonal bipyramidal structure of this chromophore is distorted toward a tetrahedron in that the cobalt ion lies below the  $\text{P}_3$  plane, the  $\text{N-Co-P}$  angle is less than  $90^\circ$  and the  $\text{Co-N}$  distance is larger than usual.<sup>7,8</sup> The polarized spectra on single crystals of this compound have been obtained with a Unicam SP 700 spectrophotometer<sup>9</sup> and are reported in Figure 1 together with the reflectance spectrum.<sup>10</sup> The reflectance spectrum of the  $[\text{CoBr}(\text{Me}_6\text{tren})]\text{Br}$  complex is also reported for comparative purposes.<sup>4</sup>

The spectrum of the former complex consists of four well resolved bands at 5.7, 10.0, 13.3, and 17.5 kK.<sup>6</sup> Their polarization is quite clear. The bands at 5.3, 9.6, and 17.2 kK are polarized in the same direction while the band at 13.0 kK is polarized perpendicularly.<sup>10</sup> By comparison with the calculated polarization, the only possible assignment is that shown in Table I. Given the close similarity of this spectrum with that of the  $\text{CoN}_4\text{Br}$  chromophore, the assignment of the bands for this complex must also be the same. This

TABLE I

Polarized spectral data for the  $[\text{CoBr}(\text{NP}_3)]\text{PF}_6$  compound

Absorption maxima (kK)	Transition	Expected Polarization	Relative direction of polarization
5.7	$^4A_2(\text{F}) \rightarrow ^4E(\text{F})$	xy	b
10.0	$\rightarrow ^4E(\text{F})$	xy	b
13.3	$\rightarrow ^4A_2(\text{P})$	z	a
17.5	$\rightarrow ^4E(\text{P})$	xy	b

supports the assignment proposed by Ciampolini and Bertini.<sup>2</sup>

The most significant difference between the spectra of the chromophores  $\text{CoN}_4\text{Br}$  and  $\text{CoNP}_3\text{Br}$  is a red shift of the bands of the latter complex in respect of those of the former one. This is well accounted for by the tetrahedral distortion of the  $\text{CoNP}_3\text{Br}$  chromophore. In fact tetrahedral complexes with  $C_{3v}$  symmetry are known to have a compressed  $d-d$  spectrum (until 16 kK) and the bands have been found to have the same polarization properties.<sup>11</sup>

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